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The range of activities cover the content of Topic C4 (Stoichiometry) of the CIE IGCSE Combined Science specification, for examination in June and November 2020 and 2021. The aim was to cover as much of the content as possible but the following topics have received particular attention:

CIE IGCSE Combined Science C4 REVISION (Stoichiometry

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GCSE SCIENCE REVISION

CIE iGCSE. Edexcel GCSE. Edexcel iGCSE. OCR GCSE. GCSE Chemistry Revision. Combined Science. GCSE Combined Science is the study of one or more of biology, chemistry and physics. Exam boards: AQA GCSE. Edexcel GCSE. OCR GCSE. GCSE Combined Science Revision. Computer Science. GCSE Computer Science is the study of algorithms, programming, computer systems and networks.

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Try the GCSE Bitesize website. The AQA syllabus is the most similar to Cambridge IGCSE. Try the S-cool websites for GCSE Biology, Chemistry or Physics notes and revision questions; Doc Brown has a great website. Here are links to the Biology, Chemistry and Physics sections.; The Cyberphysics website has information for GCSE and A level Physics students. ...

IGCSE - StampScience

Edexcel GCSE Combined Science past exam papers (9-1) and marking schemes, the past papers are free to download for you to use as practice for your exams.

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O Level Science Combined 5129 Past Papers 2019 June & Nov ...

These revision notes were made with the assumption that you

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have already studied the syllabus. So I tried to make these notes as concise as possible, but this does not mean there are any information missing. I went strictly off the syllabus, including everything you need to know for the IGCSE Double Award Coordinated Science course.

IGCSE Coordinated Science - Revision Cheat Sheets - CIE

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Complete AQA GCSE Combined Science: Synergy Past Papers
Combined Science: Synergy is part of our science suite, developed with teachers to inspire and challenge student of all abilities and aspirations. (See also GCSE Combined Science: Trilogy). Synergy is a double award and worth two GCSEs. It is assessed by four, 1 hour and 45 minute exams. [...]

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1, 2 and 3, the number of electrons lost is the same as the group number. A sodium atom loses one electron to form a sodium ion
Forming negative ions

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